2528/204 2922/204 ENVIRONMENTAL ANALYTICAL TECHNIQUES AND LABORATORY MANAGEMENT June/July 2018 Time: 3 hours



THE KENYA NATIONAL EXAMINATIONS COUNCIL

DIPLOMA IN ENVIRONMENTAL SCIENCE AND TECHNOLOGY

MODULE II

ENVIRONMENTAL ANALYTICAL TECHNIQUES AND LABORATORY MANAGEMENT

3 hours

INSTRUCTIONS TO CANDIDATES

You should have the following for this examination: Answer booklet;

Non-programmable scientific calculator.

This paper consists of TWO sections; A and B.

Answer ALL the questions in section A and any THREE questions from section B in the answer booklet provided,

Each question in section A carries 4 marks while each question in section B carries 20 marks.

Maximum marks for each part of a question are as shown.

Candidates should answer the questions in English.

This paper consists of 4 printed pages.

Candidates should check the question paper to ascertain that all the pages are printed as indicated and that no questions are missing.

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SECTION A (40 marks)

20(1-1/4)(1-34)

Answer ALL the questions in this section.

1.	Nam	(4 marks)			
2.	Outl	Outline the basic principles of steam distillation as a separation technique.			
3.	(a)	Define 'sublimation'.	(1 mark)		
	(b)	Describe how a mixture of ammonium chloride and sodium chloride ca a chemistry laboratory.	n be separated in (3 marks)		
4.	(a)	State the partition law.	(1 mark)		
	(b)	List three causes of deviation from the partition law,	(3 marks)		
5.	(a)	State two factors that determine efficiency in column chromatography.	(2 marks)		
	(b)	A column of length 120 cm has 2500 theoretical plates. Calculate the p this column.	late height in (2 marks)		
6.	Defit	ne the following terms as used in acid-base titrations:			
	(a)	alkalimetry;	(2 marks)		
	(b)	aliquot.	(2 marks)		
7.	(a)	Define 'management'.	(1 mark)		
	(b)	State any three roles of management.	(3 marks)		
8.	Outli	ine the control process of management.	(4 marks)		
9.	List	(4 marks)			
10.	(a)	State two types of planning.	(1 mark)		
	(b)	Describe the process of receiving goods in the laboratory.	(3 marks)		

SECTION B (60 marks)

Answer any THREE questions from this section.

- 11. (a) (i) Describe the preparation of filter pulp.
 - (ii) Explain two uses of filter pulp. (4 marks)
 - (b) Draw a labelled diagram of vacuum filtration in operation. (6 marks)
 - (c) State four advantages and two disadvantages of vacuum filtration as a separation method. (6 marks)
- 12. (a) (i) Define 'a primary standard' as used in volumetric analysis. (1 mark)
 - (ii) Give four examples of substances used as primary standards in:
 - I. alkalimetry; (2 marks)
 II. acidimetry; (2 marks)
 III. redox titrations. (4 marks)
 - (b) 5.804 g of a calcium ore (limestone) was dissolved in 100 cm³ of 1M HCl. The excess HCl was back-titrated with 8.3 cm³ of 0.05 M NaOH to reach the phenolphthalein end-point. Determine the percentage purity of the ore (Ca=40, C=12, O=16).

(11 marks)

- 13. (a) Explain five characteristics of a solvent used in solvent extraction. (10 marks)
 - (b) Table I shows the results obtained in an experiment to determine the partition coefficient of ethanol acid between chloroform and water.

Table I

Concentration (C _i) in chloro- form in mol.d ³	0.280	0.320	0.430	0.560	0.610
Concentration (C ₂) in water in mol.d ⁻³	0.012544			0.050176 5-02	

Use graphical method to determine the partition coefficient K_D of ethanol acid between chloroform and water. (10 marks)

14. Distinguish between proforma invoice and quotation. (a) (4 marks) (b) State any three disadvantages of tendering. (3 marks) (c) State two purposes of stock taking. (2 marks) Describe the alphabetical method of filing. (d) (11 marks) 15. Explain the essential qualities of a successful manager. (a) (8 marks)

(b) Discuss the process of dispute resolution between the employer and employee in an organisation. (12 marks)

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